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### Dr. Laurie S. Starkey, Cal Poly Pomona

#### Chapter 18, Aromatic Reactions (Ch 17/18 Part 2) - Practice Problems

Which of the following represents the molecular orbital (MO) diagram for the compound shown?



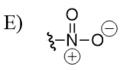
C) 
$$\frac{\pi^*}{\frac{1}{2}} \frac{\pi^*}{\pi}$$

Provide a mechanism for the following reaction.

$$\begin{array}{c|c}
\hline
OH \\
Br_2 \\
\hline
H_2O
\end{array}$$

$$\begin{array}{c}
OH \\
Br
\end{array}$$

- What effect on reactivity toward the Electrophilic Aromatic Substitution reaction would you expect if an electron-withdrawing group was placed on the aromatic ring?
- For the Electrophilic Aromatic Substitution reaction, which of the following does NOT act as an **electron-withdrawing group**?



## Categorize each of the following groups as an **ortho/para director** or a **meta director**.

Arrange the molecules shown above in order of INCREASING reactivity toward electrophilic aromatic substitution, from least reactive to most reactive?

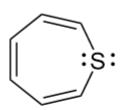
Predict the major product of the following nitration reaction and explain the regiochemistry.

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$$O$$
 $CH_3$ 
 $HNO_3$ 
 $H_2SO_4$ 

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Which of the following represents the molecular orbital (MO) diagram for the compound shown?



E) 
$$-\pi^*$$
 $\frac{1}{4}$ 
 $\frac{1}{4}$ 
 $\pi$ 

Provide a mechanism for the 2 following reaction.



- What effect on reactivity toward the Electrophilic Aromatic Substitution reaction would you expect if an **electron-withdrawing group** was placed on the aromatic ring?
  - **A)** I would expect the ring to be MORE reactive, because the ring is acting as a nucleophile.
  - **B)** I would expect the ring to be LESS reactive, because the ring is acting as a nucleophile.
  - C) I would expect the ring to be MORE reactive, because the ring is acting as an electrophile.
  - **D)** I would expect the ring to be LESS reactive, because the ring is acting as an electrophile.
  - **E)** I would expect NO effect on the reactivity of the ring (steric hindrance is the most important factor).
  - For the Electrophilic Aromatic Substitution reaction, which of the following does NOT act as an electron-withdrawing group?

A) 
$$\xi - C \equiv N$$

$$CH_3$$
B)  $\xi - C = CH_3$ 

$$E) O$$

$$\xi - NH_3$$

$$\xi - NH_3$$

$$E = 0$$

$$\xi - NH_3$$

5 Categorize each of the following groups as an ortho/para director or a meta director.

	}—СН₃	ξ—Br	ξ−CO₂Et	ξ−OCH₃
A)	o/p	meta	o/p	meta
B)	o/p	o/p	meta	meta
<b>C</b> )	meta	meta	meta	o/p
D)	o/p	o/p	meta	o/p
E)	meta	meta	o/p	meta

6 Categorize each of the following groups as an ortho/para director or a meta director.

	र्रु─NH <sub>2</sub>	र्}─NO2	ξ—NH₃	\$ N (
A)	o/p	meta	meta	o/p
B)	o/p	meta	meta	meta
C)	meta	o/p	meta	meta
D)	o/p	o/p	meta	o/p
E)	meta	meta	o/p	meta

In which of the following are the molecules shown above listed in order of INCREASING reactivity toward electrophilic aromatic substitution, from least reactive to most reactive?

- A) 2 < 4 < 1 < 3
- B) 3 < 2 < 4 < 1
- C) 3 < 4 < 2 < 1
- D) 4 < 2 < 1 < 3
- E) 4 < 3 < 1 < 2

### Predict the major product.

A) 
$$E^{+}$$

CH<sub>3</sub>
 $E^{+}$ 

C)

 $E^{+}$ 

CH<sub>3</sub>
 $E^{+}$ 

CH<sub>3</sub>
 $E^{+}$ 

CH<sub>3</sub>
 $E^{+}$ 

CH<sub>3</sub>
 $E^{+}$ 

CH<sub>2</sub>
 $E^{-}$ 

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CH<sub>3</sub>
 $E^{-}$ 

CH<sub>2</sub>
 $E^{-}$ 

CH<sub>3</sub>

Predict the major product of the following nitration reaction and explain the regiochemistry.

$$O \longrightarrow CH_3$$
 $O \longrightarrow HNO_3$ 
 $O \longrightarrow H_2SO_4$ 
 $O \longrightarrow O$ 

A) 
$$O_2N$$
  $O$   $CH_3$ 

(because -NO<sub>2</sub> is a meta director)

B) 
$$O_2N$$
  $O_2N$ 

(because -NO<sub>2</sub> is an ortho/para director)

C) 
$$O_2N$$
  $O$   $CH_3$ 

(because -OAc is a meta director)

D) 
$$O_2N$$
  $O_2N$ 

(because -OAc is an ortho/para director)