Identify the kinetic and thermodynamic products, and determine which would be favored in hot temperatures and which is favored in cold temperatures. Explain.

$$\frac{\text{HBr}}{\text{(1 equiv.)}} \qquad \frac{\text{Br}}{\mathbf{X}} \qquad + \qquad \frac{\text{Br}}{\mathbf{Y}}$$

Which of the statements below is NOT true?

- A) X is favored at high temperatures, because it is the more stable product.
- B) Y is formed faster because it comes from the more stable carbocation intermediate.
- C) Y is the kinetic product.
- D) X is the thermodynamic product.
- E) None of the above (all statements are true).