What is the name of the compound shown?



- A) Oxazole
- B) Oxobenzene
- C) Pyrrole
- D) Furan
- E) Tetrahydrofuran (THF)

Is the given compound aromatic?

- A) Yes, because it cyclic, with 6 electrons in contiguous p orbitals.
- B) Yes, because it cyclic, with 8 electrons in contiguous p orbitals.
- C) No, because 8 electrons does not satisfy Hückel's Rule.
- D) No, because it is not planar.