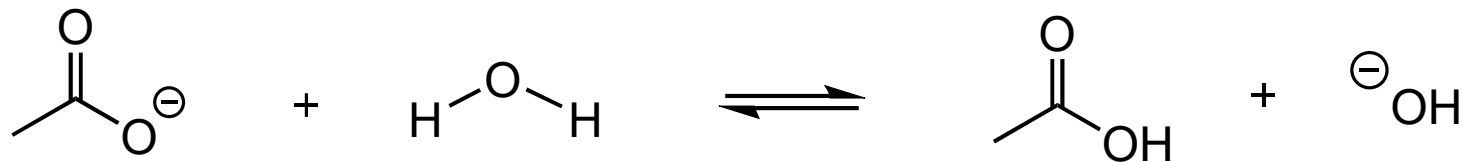


Provide curved arrows for the following proton-transfer reaction.
Determine the direction of the equilibrium (forward or reverse favored?).



- A) Forward, because acetate is the weaker base.
- B) Forward, because hydroxide is the weaker base.
- C) Reverse, because acetate is the stronger base.
- D) Reverse, because hydroxide is the stronger base.