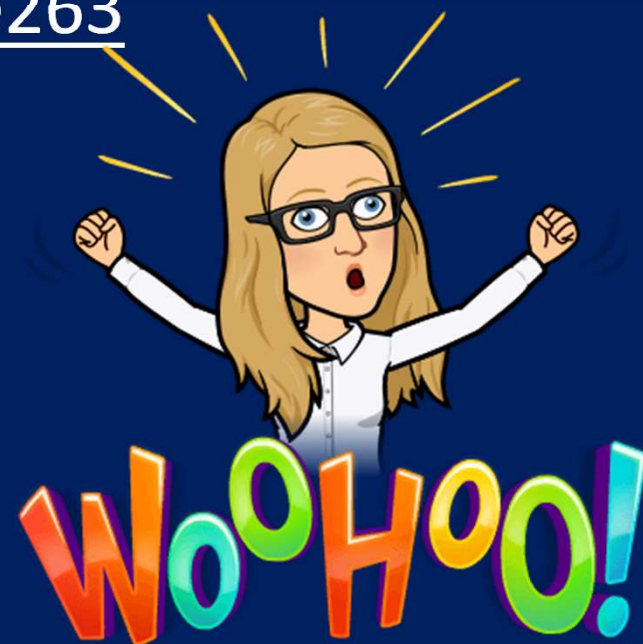




For clicker question voting, go to:
<https://pollev.com/lauriestarke263>

Dr. Laurie S. Starkey
Cal Poly Pomona



CHM 3150 Organic Chemistry II
Announcements 12/4/25 – LAST DAY!!

Today's Topic:

Chapter 22 Amines

Chapter 22

- ✓ Watch
- ✓ Read
- ✓ Practice

Step 1

- Read Klein 22.1, 22.2, 22.3 for nomenclature and properties of amines
- Note: we have already covered diazonium salts (ArN_2^+), sections 22.10 and 22.11
- Watch flipped lecture
- Work through **SkillBuilder 22.1** and **Conceptual Checkpoints** (problems 22.1 - 22.9, 22.23)
- ~~FYT (will not be on the final).~~
- Klein 22.9 **Hofmann Elimination**

Amines

35 minutes

skeleton notes pages 22-1 to 22-4

Step 2 - *OPTIONAL* *GOOD FOR FINAL EXAM REVIEW*

- Read Klein 22.4 - 22.7 **Preparation of Amines**
- Watch flipped lecture
- Work through **SkillBuilder 16.3** (problems 21.14 - 21.28)
- Suggested textbook problems: 1-67 (skip 5, 19-28, 41, 47-56, 59, 60c, 65)
- **Diels-Alder homework** (submit to [Gradescope](#)) and [answer key](#)

Synthesis of Amines

10 minutes

skeleton notes page 22-5

Educator: Ch. 22 Amines

Amines	34:58
Intro	0:00
Amines: Properties and Reactivity	0:04
Compare Amines to Alcohols	0:05
Amines: Lower Boiling Point than ROH	0:55
1) RNH ₂ Has Lower Boiling Point than ROH	0:56
Amines: Better Nu: Than ROH	2:22
2) RNH ₂ is a Better Nucleophile than ROH Example 1	2:23
RNH ₂ is a Better Nucleophile than ROH Example 2	3:08
Amines: Better Nu: than ROH	3:47
Example	3:48
Amines are Good Bases	
3) RNH ₂ is a Good Base	
Amines are Good Bases	
Example 1	
Example 2: Amino Acid	
Alkyl vs. Aryl Amines	
Example: Which is Strongest Base?	
Alkyl vs. Aryl Amines	
Verify by Comparing Conjugate Acids	
Reaction of Amines	
Reaction with Ketone/Aldehyde: 1° Amine (RNH ₂)	
Reaction of Amines	
Reaction with Ketone/Aldehyde: 2° Amine (R ₂ NH)	
Use of Enamine: Synthetic Equivalent of Enolate	
Use of Enamine: Synthetic Equivalent of Enolate	

Section 10: Properties & Reactions of Amines

Synthesis of Amines	46:05
Synthesis of Amines	46:06
Gabriel Synthesis of Amines	47:57
Gabriel Synthesis of Amines	47:58
Amines by SN2 with Azide Nu:	49:50
Amines by SN2 with Azide Nu:	49:51
Amines by SN2 with Cyanide Nu:	50:31
Amines by SN2 with Cyanide Nu:	50:32
Amines by Reduction of Amides	51:30
Amines by Reduction of Amides	51:31
Reductive Amination of Ketones/Aldehydes	52:42
Reductive Amination of Ketones/Aldehydes	52:43
Example : Synthesis of an Amine	53:47
Example 1: Synthesis of an Amine	53:48
Example 2: Synthesis of an Amine	56:16

Section 12: Synthesis of Amines

Final Exam covers Chapters 1-22!

CHM 3150 Organic Chemistry II, Dr. Laurie S. Starkey, Fall 2025					
<i>Tentative Schedule (Chapter and Worksheet #)</i>					
Week	Mon	Tues	Wed	Thurs	Fri
	11/24	11/25	11/26	11/27	11/28
14		Exam III		Holiday	Holiday
	12/1	12/2	12/3	12/4	12/5
15		Ch. 16 #2		Ch. 22 #1	
Finals (section)	12/8	Tue. 12/9 1:00–2:50pm (02/1pm) 5:00–6:50 pm (03/5:30)	12/10	Thurs. 12/11 3:00–4:50 pm (01/4pm)	12/12



110-minute multiple-choice ACS exam
In-person on 12/9 or 12/11
 Scantron and scrap paper will be provided!

Final Exam

Section 02 (1 pm class in 24D-1224): **Tuesday 12/9 1:00–2:50 pm**
 Section 03 (5:30 pm class in 24A-1403): **Tuesday 12/9 5:00–6:50 pm**
 Section 01 (4 pm class in 15-1823): **Thursday 12/11 3:00–4:50 pm**

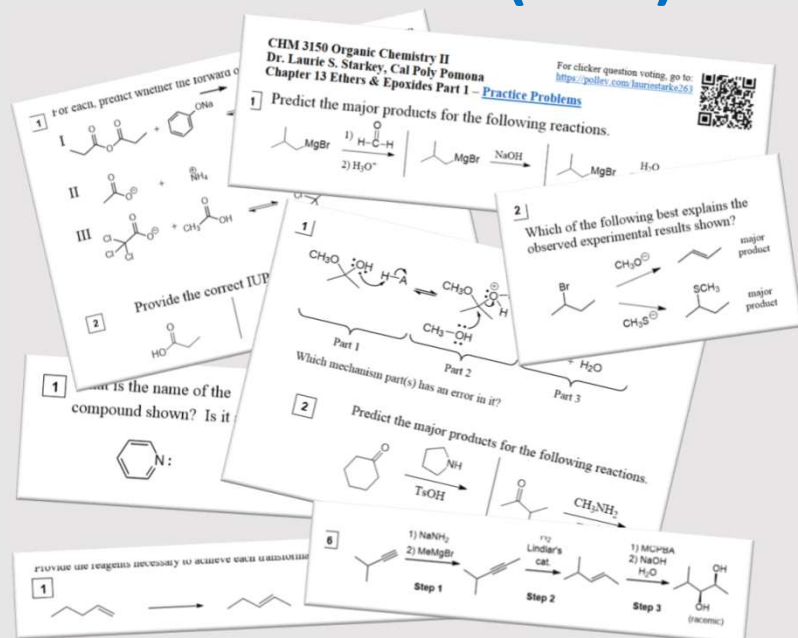
Review session: Sunday 12/6, 8 pm Worksheet

(review covers mostly CHM 3140 topics, see Week 1-15 worksheets above for CHM 3150 topics!)

Final Exam Review

CHM 3150 Exams 1/2/3 & Exam Reviews 1/2/3

CHM 3150 Homework & Worksheets (x27!)

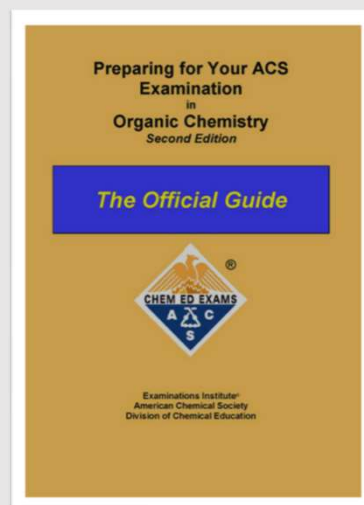
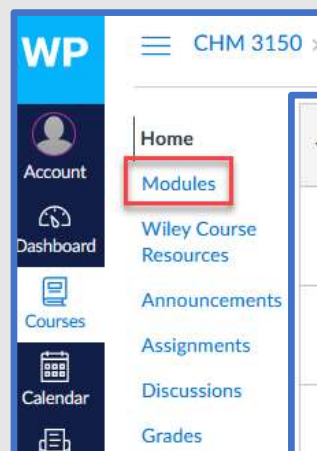


Resources on Course Homepage

ACS Final Exam (Ch. 1-22 topics):

- [ACS Exam Practice handout](#) provided by the ACS and [answer key](#) and [detailed solutions](#)
- [Practice multiple-choice problems](#) written by Dr. Reusch
- All Functional Groups Nomenclature [Review Problems](#) and [answer key](#)
- Stereochemistry [practice problems](#) and [answer key](#)
- [IR and NMR lectures and handouts](#)

WileyPLUS Assignments

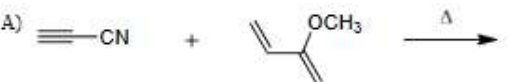



ACS Study Guide (library reserve)

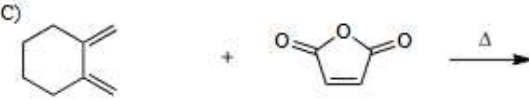
Last “Free Red Ink” Homework: Diels-Alder

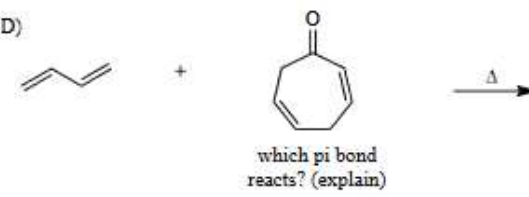
*No EOC-SkillBuilder
credit for Ch 16/22*

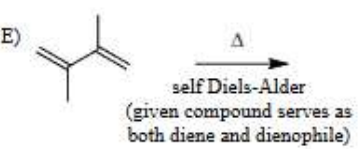
Organic Chemistry II CHM 3150, Dr. Laurie S. Starkey, Cal Poly Pomona
Diels-Alder Homework
Name: _____ Section (day/time): _____

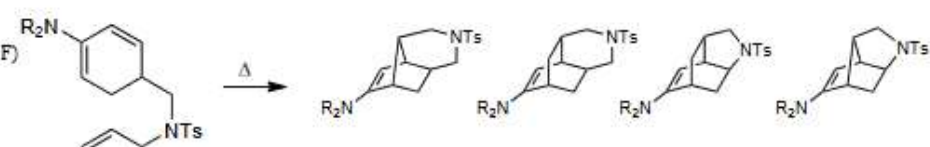
A) 

B) 

C) 

D) 

E) 

F) 
select the correct product
(show your work!)

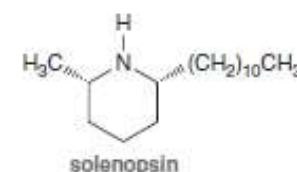
An Amino Neurotoxin Beaten by its Basicity



WorldLinks | Chemical Warfare Among Ants

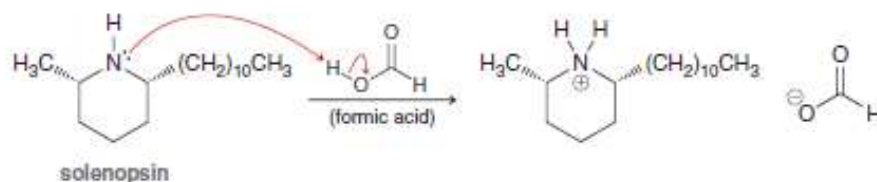
Humans are not the first species to use chemical warfare. Many ant species can inject formic acid when they bite, and the formic acid greatly increases the pain of the bite (*formica* is the Latin word for ant). Fire ants, a genus of aggressive ants that invaded the southern United States from Argentina in the 1920s, have amplified the level of chemical warfare by making an even more toxic compound, solenopsin, which is a cyclic amine (shown here).

Solenopsin, a potent neurotoxin, is a hydrophobic molecule that dissolves through the waxy cuticle layer of insects. Fire ants can squirt solenopsin onto enemy ants and kill them with their toxin. Tawny crazy ants, another invasive species of aggressive ants from Argentina, have invaded the southern US more recently and have been taking over territory inhabited by fire ants. In battles between fire ants and crazy ants, the crazy ants do get sprayed with the fire ants' neurotoxin, but they retreat and secrete formic acid from the tip of their abdomen and groom themselves with formic acid. Afterwards, they return to battle, seemingly unaffected by the exposure



A creative rendition of a crazy ant, armed and ready to battle fire ants.

to the fire ants' neurotoxin. How does the formic acid block the action of solenopsin? Although there is no definitive answer yet, the simplest explanation is that formic acid chemically neutralizes the amine group of solenopsin, forming an ionic ammonium ion salt that is not permeable/soluble in the ant's hydrophobic cuticle.



As a result, the ionic form of solenopsin never gets inside the crazy ants. Virtually unaffected by the neurotoxin, the crazy ants are able to continue battling the fire ants.

The phenomenon of one ant taking over the territory of another ant had never been observed in the US before the invasion of the crazy ants. This is a great example of how chemistry is involved in animal behavior and ecology.

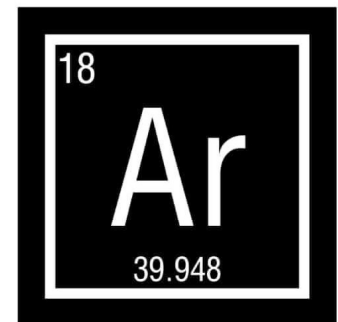


Thank you for your dedication & hard work this semester!

- It has been a pleasure working with you!
- Good luck on finals & have a great winter break!



I MISS YOU WHEN YOU



All I want for
Christmas
is ^{235}U

