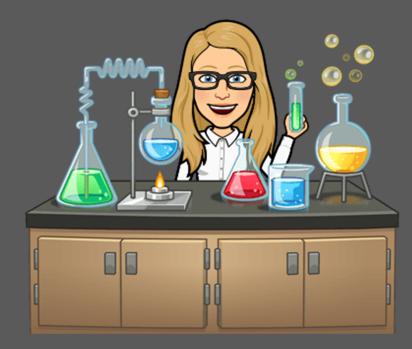
### For clicker question voting, go to: <a href="https://pollev.com/lauriestarke263">https://pollev.com/lauriestarke263</a>





Dr. Laurie S. Starkey Cal Poly Pomona

# CHM 3150 Organic Chemistry II 9/25/25

# Today's Topic: Wittig Reaction & Formation of Acetals (Chapter 19, Part 2)

#### **Chapter 19**

- ✓ Watch
- ✓ Read
- ✓ Practice

#### Step 2

#### Read

- Read Klein 19.6 Reaction with Carbon Nucleophiles
- Work through SkillBuilder 19.2 and Conceptual Checkpoint 19.15ab
- Read Klein 19.7 Reactions with Oxygen Nucleophiles (hold off on Protecting Groups until later)
- Work through SkillBuilder 19.3 and Conceptual Checkpoints 19.20ab, 19.21, 19.22.
- Note: for more sugar examples, see SkillBuilder 24.1 and Chapter 24 problems 24.9, 24.10, 24.11.
- Read Klein 19.9 Hydrolysis of Acetals

- Part 2a Wittig

  13 minutes

  skeleton notes pages 19-4 to 19-5
- Part 2b Acetal Fermation

  29 minutes

  skeleton notes pages 19-5 to 19-7
- Part 2c Acetal Hydrolyst
   10 minutes
   skeleton notes page 10-7

**Practice** 

Watch

# Flipped Lectures: Wittig Reaction & Formation of Acetals

		vvittig Reaction			
		Wittig Reaction: A R	Resonance-Stabilized Carl	banion (Nu:)	
		Wittig Reaction: Me	echanism		
Reaction with Oxygen Nu:		Preparation of Wittig Re Two Steps From RX	_		Wittig Rxn
Addition of H₂O		Example: Predict			Tittig Total
Exception: Formaldehyde is 99% Hydrate in		Wittig Retrosynthesis			
Exception: Hydrate is Favored if Partial Posi	itive Near Carbonyl	Wittig Retrosynthes			
Reaction with Oxygen Nu:			515		
Addition of ROH		Synthesis			
TsOH: Tosic Acid			58:28		
Addition of ROH Cont.			59:09		
Example			61:43	E	
Predict			61:44		
Mechanism			63:08		HELLO
Mechanism for Acetal Formation	_		64:10		
Mechanism for Acetal Formation	Oxyge	en Nu:	64:11		my name is
What is a CTI?			75:04		Wittig
Tetrahedral Intermediate	Ace	tais	75:05		
Charged Tetrahedral Intermediate			75:45		reagent
CTI: Acid-cat			76:10		
CTI: Base-cat			77:01		H <sub>a</sub> C=PPh <sub>a</sub>
Acetals & Cyclic Acetals			77:49		1120-11113

77:50

78:46

Overall

Cyclic Acetals

37:39 37:40 39:51 41:58

45:02 46:19

48:09

#### Ch. 19 Free Red Ink Homework x2

#### California State Polytechnic University, Pomona

Organic Chemistry II, CHM 3150, Dr. Laurie S. Starkey Alcohol Synthesis Homework

ction: (day/time

A) Suggest **two** possible syntheses for the following target molecule (**TM**), starting with any **aldehyde** (start each synthesis with a different aldehyde). It may help to first consider a retrosynthesis of the TM.

Hint! Possible aldehyde starting materials:

CHM 3150 Organic Chemistry II, Dr. Laurie S. Starkey, Cal Poly Pomona
Acetal Mechanism Homework

Name: \_\_\_\_\_\_ Section: \_\_\_\_\_ (day/time)

STOP! Before working on this homework, review Mechanism 19.6 and work on SkillBuilder 19.3. Provide complete mechanisms for both the formation of the cyclic acetal and then the hydrolysis of the acetal. Pay close attention to details, including lone pairs, formal charges and the use of curved arrows.

Acetal formation mechanism:

#### All Course Points go through Gradescope

Exam I assignments*		Exam II assignments*		Exam III assignments*		
WileyPLUS/textbook		WileyPLUS/textbook		WileyPLUS/textbook		
SkillBuild/EOC Ch. 11	10	SkillBuild/EOC Ch. 19	10	SkillBuild/EOC Ch. 21	10	
SkillBuild/EOC Ch. 12	10	SkillBuild/EOC Ch. 20	10	SkillBuild/EOC Ch. 17	5	
SkillBuild/EOC Ch. 13	10			SkillBuild/EOC Ch. 18	10	Homework
						Course Points
"Free Red Ink Homework"		"Free Red Ink Homework"		"Free Red Ink Homework"	"Free Red Ink Homework"	
Hmwk - Ch. 11 EOC/Review	5	Hmwk ROH synthesis	5	Hmwk 8 Messy Aldol	5	125 (25%)
Hmwk - Grignard	5	Hmwk acetal	5	Hmwk 9 biphenyl EAS	5	*assignments due no later
Hmwk - epoxide	5	Hmwk hydrolysis	5	Hmwk 10 FC alkylation	5	than date of each exam
		Hmwk - ArOH acidity	5			
		-				
	45		40		40	
Friday Fives (4 pts each)	52	OLC reports (1 pt each)	10	Exam Wrappers (4 pts each, drop one)	8	Study/Reflection Course Points
	52		10			Course Follits
(13+ earns max. credit)		(10+ reports earns max. credit)		Exam Corrections	8	75 (450/)
				(4 pts each, drop one)		75 (15%)
Midterm Exams x3	)	(100 pts each, drop lowest)		Final Exam		Midterms: 200 (40%) Final: 100 (20%)



#### c&en

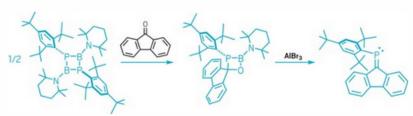
REAGENTS

#### Phospha-bora-Wittig reaction makes its debut

A new variant of the classic Wittig reaction offers a route to phosphaalkenes

by Mark Peplow, special to C&EN

September 21, 2021



Credit: Adapted from J. Am. Chem. So.

A new phosphaborene dimer (left) reacts with a ketone to form a 4-mem similar to the intermediate formed using classic Wittig reagents. The int a phosphalkene (right), a molecule with a carbon-phosphorus double bo

# Got Research? Synthetic Techniques are Continuously Being Developed

utting a new twist on an old favorite of organic chemistry, researchers have unveiled the phospha-bora-Wittig reaction. It offers a relatively straightforward way to prepare phosphaalkenes, which are used as ligands in transition-metal catalysts and as polymer building blocks. But its inventors, led by Michael J. Cowley of the University of Edinburgh, say that it might also help to expand the scope of the original Wittig reaction so that it works with a wider range of molecules (J. Am. Chem. Soc. 2021, DOI: 10.1021/jacs.1c06228).