Given the pK_a values shown, in which direction does the equilibrium lie? Explain briefly. (next, try it without pK_a !)

$$^{\oplus}$$
NH₄ + H₂O \longrightarrow NH₃ + H₃O $^{\oplus}$
p K_a 9 16 36 -2

- A) Forward, because the combined pK_a 25 is lower than 34.
- B) Reverse, because the combined pK_a 25 is lower than 34.
- C) Forward, because -2 is lower than 9.
- D) Reverse, because -2 is lower than 9.
- E) Neither, because the charges are balanced.