Which of the following is the FASTER reaction? Explain briefly.

I
$$H_2O$$
 OH

II H_2O OH

- A) I is faster because this is more stable:
- B) II is faster because this is less stable:
- C) I is faster because LG is allylic
- D) II is faster because LG has less sterics
- E) II is faster because the LG is tertiary