Which of the following is the FASTER reaction? Explain briefly.

I
$$\frac{\text{NaN}_3}{\text{acetone}}$$
 $\frac{\text{NaN}_3}{\text{NaN}_3}$ $\frac{\text{NaN}_3}{\text{acetone}}$ $\frac{\text{NaN}_3}{\text{Nacetone}}$

- A) I is faster because this is more stable:
- B) II is faster because this is less stable:
- C) I is faster because LG has less sterics
- D) II is faster because LG has less sterics