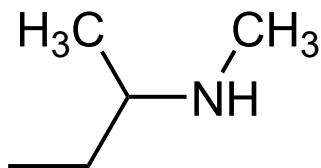
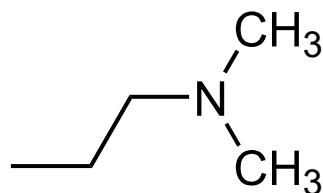


Which has the higher boiling point/bp (A, B or neither)?
Explain briefly.



A



B

- A) Neither bp is higher because they are both amines.
- B) A has the higher bp because it is less branched
(A has more van der Waals attractions).
- C) A has the higher bp because it has a hydrogen bond.
- D) A has the higher bp because the N-H bond is strong.
- E) A has the higher bp because the N-H group can form hydrogen bonds.