

CHM 3140 Organic Chemistry I, Cal Poly Pomona, Dr. Laurie S. Starkey

Chapter 9 (Klein) Summary: Reactions of Alkynes

I. Structure, Properties and Nomenclature of Alkynes (9.1, 9.2) **SkillBuilder 9.1**

II. Alkyne preparation by E2/dehydrohalogenation

- i) – HX twice w/strong base (9.4)

III. Reductions of alkynes (9.5)

- i) H₂/catalyst (Pd, Ni, Pt, RhCl[PR₃]₃) adds 2 equiv. to give alkane
- ii) H₂/poisoned cat. (Lindlar's catalyst) adds 1 equiv. to give cis alkene
- iii) Na/NH₃ (dissolving metal reduction) converts internal alkyne to a trans alkene

IV. Addition reactions

- i) Br₂ (adds 2 equivalents) (9.8)
- ii) HBr (adds 2 equivalents, both Markovnikov) (9.6)
- iii) Hydration of alkynes (9.7) **SkillBuilders 9.3, 9.4**
 - A) Markovnikov addition of H-OH (1 equiv.) to give enol
 - B) tautomerization of enol to ketone/aldehyde (1st step protonate; 2nd step deprotonate)

V. Ozonolysis of alkynes (9.9)

- i) cleave C≡C triple bond to give 2 carboxylic acids

VI. Alkyne synthesis by alkylation reaction (9.10)

- i) deprotonation of alkynes (9.3)
 - A) acidity of alkynes (RC≡CH + NaNH₂ → RC≡CNa) **SkillBuilder 9.2**
 - B) RC≡CNa can be a nucleophile (S_N2 on MeI or primary R'X) **SkillBuilder 9.5**

VII. Synthesis strategies (9.11) **SkillBuilder 9.6**

- i) Are there any changes to the carbon skeleton? (use RC≡CNa nucleophile + R'X)
- ii) Are there any changes to the identity/position of functional group?
- iii) Make logical disconnections to develop a retrosynthetic plan.