Dr. Starkey, CHM 3140 Organic Chem. I, Cal Poly Pomona Chapter 7 Part 2 – Elimination Rxns (<u>Ch 7 Worksheet #3</u>)

For clicker question voting, go to: https://pollev.com/lauriestarke263 LAURIESTARKE263 to 37607



Select the more stable alkene in each pair of compounds.

1

Arrange alkenes from most stable to least stable:

$$A$$
 B
 C
 D

3 Example: CH₃ NaOH

E2 CH₃ Br

"eliminate" or lose β-hydrogen and LG (- HBr called dehydrohalogenation)

Group work: draw a Transition State for E2 reaction given above.

4

Group work: predict the major product for the following E2 elimination reactions. Consider both regiochemistry and stereochemistry.

5

In each case, determine the mechanism (S_N2, E2 or N.R.) and predict the major product(s).

CH₃ EtONa EtOH

$$\begin{array}{c|c} & NH_3 \\ \hline & DMF \end{array}$$

8

$$\frac{t\text{-BuOK}}{t\text{-BuOH}}$$

$$Ph \xrightarrow{NaOH} H_2O$$