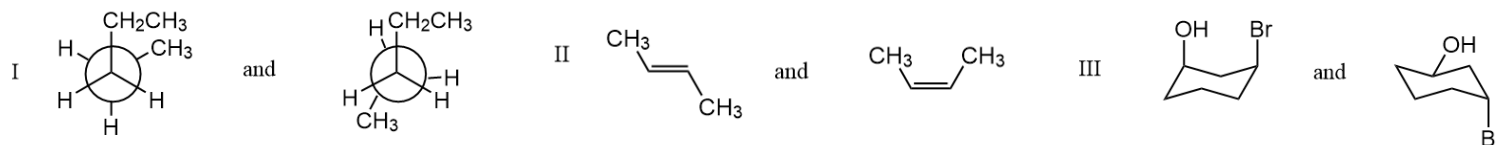


Chapter 5 Stereochemistry, Part 1 – [Practice Problems](#)

Many of these problems are from the Ch. 5 skeleton notes ([page](#)).

1

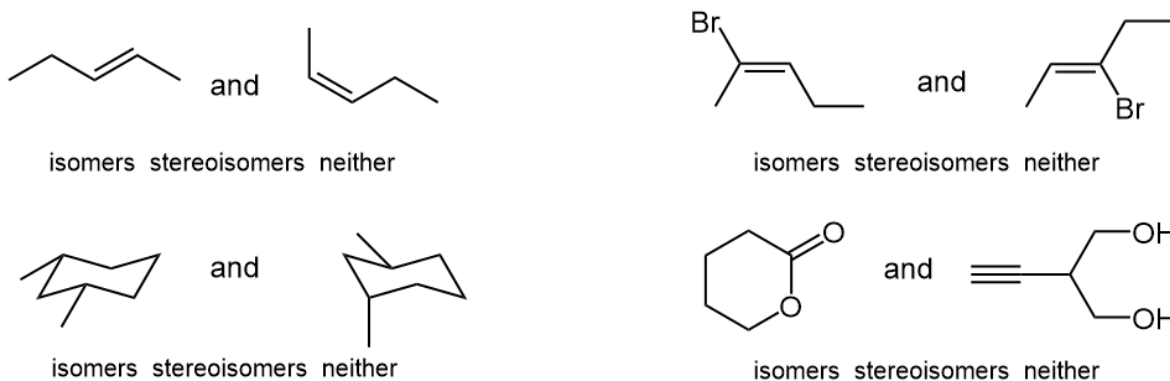
Which of the following pairs are examples of **conformers**?



2

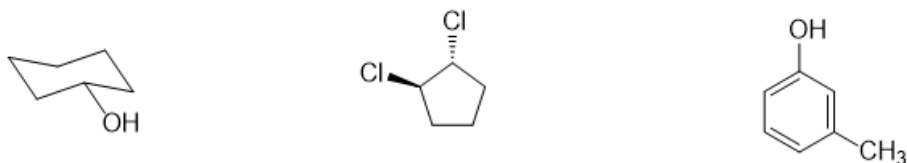
[p.5-1](#)

Group Work: categorize each pair of compounds (circle one: isomers, stereoisomers or neither).



3

Determine whether each of the given compounds is chiral or achiral.



[p.5-3](#)

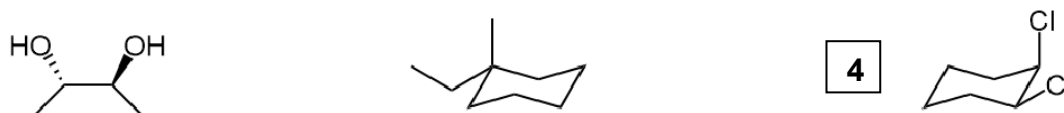
All **chiral** objects have non-superimposable mirror images. (e.g., a student desk)

Every chiral molecule has an _____

All **achiral** objects are exactly the same as their mirror images. (e.g., a chair without arms)

An achiral molecule does / does not have an enantiomer.

Determine whether each molecule is *chiral* or *achiral*, and determine whether or not it has an enantiomer. Consider redrawing chairs as hexagons.



Chiral? _____

Has an _____
enantiomer?

Chiral? _____

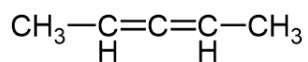
Has an _____
enantiomer?

Chiral? _____

Has an _____
enantiomer?

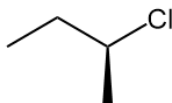
5

Use drawings to explain why the following compound is chiral.
(*hint: draw a 3D sketch and look for planes of symmetry*) (Klein 5.9)

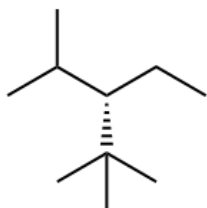


6

Give a complete name for the given compound (include stereochemistry).

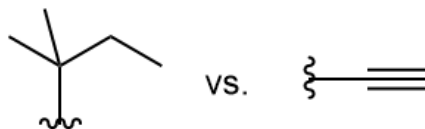
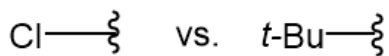


7



Group activity: which substituent would have the higher priority?

8



9

Provide a drawing for the following name:
(3*S*, 4*R*)-1,3,4-trichloro-2-(chloromethyl)heptane