California State Polytechnic University, Pomona Dr. Laurie S. Starkey, Organic Chemistry Lab, CHM 424L  Column Chromatography Quiz
1) In 1978, W. Clark Still developed "flash" column chromatography. How did it differ from the traditional columns run at that time? (what did he do differently and how was it better?)
2) What size silica gel (in μm) is ideal for flash column chromatography?
3) To what height should the column be packed with silica gel?
4) In what place(s) is sand used in the column?
5) You will be preparing your column by passing solvent through the silica gel. How do you know when it is done being packed?
6) What is an ideal flow rate of solvent?
7) What is an ideal $R_f$ value for a compound that you want to purify by flash column?
8) We will be using a solvent mixture of ethyl acetate (EtOAc) and hexane. If you wanted your spots to have a <u>higher</u> $R_f$ , how would you adjust your solvent mixture? i.e., would you add more ethyl acetate or more hexane? Explain.
9) Assuming you had a good separation between your two spots (e.g., $R_f$ 0.2 and $R_f$ 0.5 on a given TLC plate), what diameter column would you use for a 1 gram sample?

10) Your sample may be a solid or a liquid or an oil. How is it to be loaded onto the column?